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TECHNICAL REPORT ARBRL-TR-02417

APPLICATION OF CARS TO OBTAIN  
TEMPERATURE IN FLAME ENVIRONMENTS

John A. Vanderhoff  
Anthony J. Kotlar

August 1982



US ARMY ARMAMENT RESEARCH AND DEVELOPMENT COMMAND  
BALLISTIC RESEARCH LABORATORY  
ABERDEEN PROVING GROUND, MARYLAND

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20. burning) are discussed in the report.

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## I. INTRODUCTION

The use of laser light to probe combustion processes has become a very active field of research. Because of the non-perturbing nature of light and the monochromaticity of the laser, this tool is well suited for making detailed spatially and temporally resolved measurements in combustion environments. Raman techniques can be readily applied for probing combustion processes at elevated pressure since the Raman scattering processes occur on time scales less than  $10^{-12}$ s, a time short compared to the collision times.

Spontaneous Raman spectroscopy has been used to probe relatively clean flame systems<sup>1-9</sup>; however, background noise effects such as luminosity, fluorescence, and particle incandescence can obscure the spontaneous Raman signals. With the development of very high power visible lasers, non-linear optical spectroscopy became feasible. Coherent anti-Stokes Raman scattering (CARS)<sup>10-11</sup> has emerged as one of the most promising techniques for probing

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<sup>1</sup>M. Lapp and C.M. Penney, Eds., Laser Raman Gas Diagnostics, Plenum Press, New York, 1973.

<sup>2</sup>S. Lederman, "The Use of Laser Raman Diagnostics in Flow Fields and Combustion," Prog. Energy Combust. Sci. Vol. 3, pp. 1-34 1977.

<sup>3</sup>M. Lapp, "Raman Scattering Measurements of Combustion Properties," Laser Probes for Combustion Chemistry, ACS Symposium Series 134, Washington, D.C., 1980.

<sup>4</sup>J.H. Bechtel, "Temperature Measurements of the Hydroxyl Radical and Molecular Nitrogen in Premixed Laminar Flames by Laser Techniques," Applied Optics Vol. 18, p. 2100, 1979.

<sup>5</sup>M. Bridoux, M. Crunell-Cras, F. Grase, and Michel Delhaye, "Use of Multichannel Pulsed Raman Spectroscopy as a Diagnostic Technique in Flames," Combustion and Flame, Vol. 36, pp. 109-116, 1979.

<sup>6</sup>M.C. Drake and G.M. Rosenblatt, "Rotational Raman Scattering from Premixed and Diffusion Flames," Combustion and Flame, Vol. 33, pp. 179-196, 1978.

<sup>7</sup>G. Alessandretti, "Some Results on the Measurement of Temperature and Density in a Flame by Raman Spectroscopy," Optica Acta, Vol. 27, pp. 1095-1103, 1980.

<sup>8</sup>A.A. Bojarski, R.H. Barnes, and J.F. Kircher, "Flame Measurements Utilizing Raman Scattering," Combustion and Flame, Vol. 32 pp. 111-114, 1978.

<sup>9</sup>D.P. Aeschliman, J.C. Cummings, and R.A. Hill, "Raman Spectroscopic Study of a Laminar Hydrogen Diffusion Flame in Air," J. Quant. Spectrosc. Radiat. Transfer Vol. 21, pp.293-307, 1979.

<sup>10</sup>A.C. Eckbreth, P.A. Bonczyk, and J.F. Verdieck, "Laser Raman and Fluorescence Techniques for Practical Combustion Diagnostics," Appl. Spectros. Reviews, Vol. 13, pp. 15-164, 1978.



the more hostile combustion environments. Highly sooting flames<sup>12</sup>, combustors<sup>13,14</sup>, internal combustion engines<sup>15</sup>, and gun propellant flames<sup>16</sup> have been investigated using CARS. In most of these studies temperature profiles alone were obtained but in certain cases concentrations have also been measured with CARS<sup>17-19</sup>. This paper presents our results of testing CARS as a combustion diagnostic in low luminosity flames ( $\text{CH}_4/\text{air}$ ) and luminous flames ( $\text{CH}_4/\text{N}_2\text{O}$ ) and the use of a computer analysis required for interpreting CARS spectra.

## II. BACKGROUND

The first observation of the non-linear mixing process termed CARS was reported by Maker and Terhune<sup>20</sup> in 1965 and was only much later applied to combustion by Regnier and Taran<sup>21</sup> in 1973. Several review articles discuss

- <sup>11</sup>J.W. Nibler and G.V. Knighten, "Coherent Anti-Stokes Raman Spectroscopy," Raman Spectroscopy of Gases and Liquids, Topics in Current Physics, Vol. II, Springer-Verlag, New York, 1979.
- <sup>12</sup>A.C. Eckbreth, R.J. Hall and J.A. Shirley, "Investigations of Coherent Anti-Stokes Raman Spectroscopy (CARS) for Combustion Diagnostics," AIAA Paper 79-0083, 17th Aerospace Sciences Meeting, New Orleans, La., Jan. 1979.
- <sup>13</sup>A.C. Eckbreth, "CARS Thermometry in Practical Combustors," Combustion and Flame, Vol. 39, pp. 133-147, 1980.
- <sup>14</sup>G.L. Switzer, L.P. Goss, W.M. Roquemore, R.P. Bradley, P.W. Schreiber, and W.B. Roh, "Application of CARS to Simulated Practical Combustion Systems," Journal of Energy, Vol. 4, pp. 209, 1980.
- <sup>15</sup>I.A. Stenhouse, D.R. Williams, J.B. Cade, and M.P. Swords, "CARS in an Internal Combustion Engine," Applied Optics, Vol. 18, pp. 3819, 1979.
- <sup>16</sup>L.E. Harris and M.E. McIlwain, "CARS Spectroscopy of Gun Propellant Flames," Technical Report ARLCD-TR-81007, 1981.
- <sup>17</sup>A.C. Eckbreth and R.J. Hall, "CARS Concentration Sensitivity With and Without Nonresonant Background Suppression," Combustion Science and Technology, Vol. 25, pp. 175-192, 1981.
- <sup>18</sup>M. Pealat, B. Attal, and J.P.E. Taran, "CARS Diagnostics of Combustion" AIAA 80-0282, 1980.
- <sup>19</sup>L.A. Rahn, L.J. Zych, and P.L. Mattern, "Background-free CARS Studies of Carbon Monoxide in a Flame," Optical Communications, Vol. 39, pp. 249, 1979.
- <sup>20</sup>P.D. Maker and R.W. Terhune, "Study of Optical Effects Due to an Induced Polarization Third Order in the Electric Field Strength," Phys. Rev. Vol. 137, pp. A801-A818, 1965.
- <sup>21</sup>P.R. Regnier and J.P.E. Taran, "On the Possibility of Measuring Gas Concentrations by Stimulated Anti-Stokes Scattering," Appl. Phys. Lett., Vol. 23, pp. 240-242, 1973.

the CARS technique in detail<sup>11,22,23</sup> and thus only a brief description will be given here.

CARS is a four wave mixing process in which the interaction occurs through the third order non-linear susceptibility ( $\chi^{(3)}$ ) which is associated with the third power of the applied electric field. For isotropic materials, gases and liquids,  $\chi^{(3)}$  is the lowest order non-linear interaction. Two waves at a pump frequency  $\omega_1$  and one at a Stokes frequency  $\omega_2$  mix to produce a new coherent beam at the anti-Stokes frequency  $\omega_3$  which is equal to  $2\omega_1 - \omega_2$ . This is visualized through the use of virtual states as is shown in Fig. 1a. This mixing process will occur at all frequencies but is greatly enhanced when  $\omega_1 - \omega_2$  corresponds to a Raman active vibrational-rotational resonance ( $\omega_{VR}$ ). By scanning across  $\omega_2$  or using broad band radiation centered at  $\omega_2$ , a spectrum can be obtained which contains information similar to a spontaneous Raman Stokes spectrum, Fig. 1b.

In addition to energy matching the incident laser beams to obtain a resonance condition, the momentum or phase matching condition  $\Delta k = 2k_1 - k_2 - k_3 = 0$  is required. Here the momentum vector  $k$  is equal in magnitude to  $n\omega/c$  where  $n$  is the frequency dependent index of refraction and  $c$  the speed of light. Three different geometrical configurations of vectors shown in Fig. 2 can meet this condition. In the most general crossed beam case two incident laser beams with momentum vectors  $k_1$  and  $k_2$  cross at a small angle  $\theta$ . Phase matching can be made to work for samples in which the index of refraction varies with frequency. As an example phase matching can be obtained for the  $992\text{ cm}^{-1}$  vibrational transition in benzene using  $\theta \sim 1^\circ$ . For gases the index of refraction varies little with frequency and  $\theta \rightarrow 0$  resulting in a collinear arrangement of beams, Fig. 2b. This alignment has the advantage of a long interaction length thus producing large CARS signals (CARS depends on the square of the interaction length).<sup>23</sup> Disadvantages of this configuration are poor spatial resolution and unwanted CARS signal contributions. In the majority of cases the temperature of a flame is determined from the information in the CARS signal for  $N_2$ . Here contributions from room air outside of the flame can dominate the CARS signal if careful shielding of the overlapped laser beams is not used. A method of circumventing these problems and still meeting the phase matching requirement has been introduced by Eckbreth<sup>24</sup> and is termed BOXCARS. In this case the pump laser beam at  $\omega_1$  is split into two components shown in Fig. 2c and the resulting vector diagram for  $\Delta k = 0$  forms the shape of a box. High spatial resolution can be obtained with this method. Both collinear and BOXCARS

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<sup>22</sup>S.A. Akhmanov and N.I. Koroteev, "Spectroscopy of Light Scattering and Nonlinear Optics. Nonlinear-Optical Methods of Active Spectroscopy of Raman and Rayleigh Scattering," *Sov. Phys. Usp.* Vol. 20, pp. 899-936, 1977.

<sup>23</sup>W.M. Tolles, J.W. Nibler, J.R. MacDonald, and A.B. Harvey, "A Review of the Theory and Application of Coherent Anti-Stokes Raman Spectroscopy (CARS)," *Applied Spectroscopy*, Vol. 31, pp. 253-271, 1977.

<sup>24</sup>A.C. Eckbreth, "BOXCARS: Crossed-Beam Phase-Matched CARS Generation in Gases," *Appl. Phys. Lett.* Vol. 32, pp. 421-423, 1978.

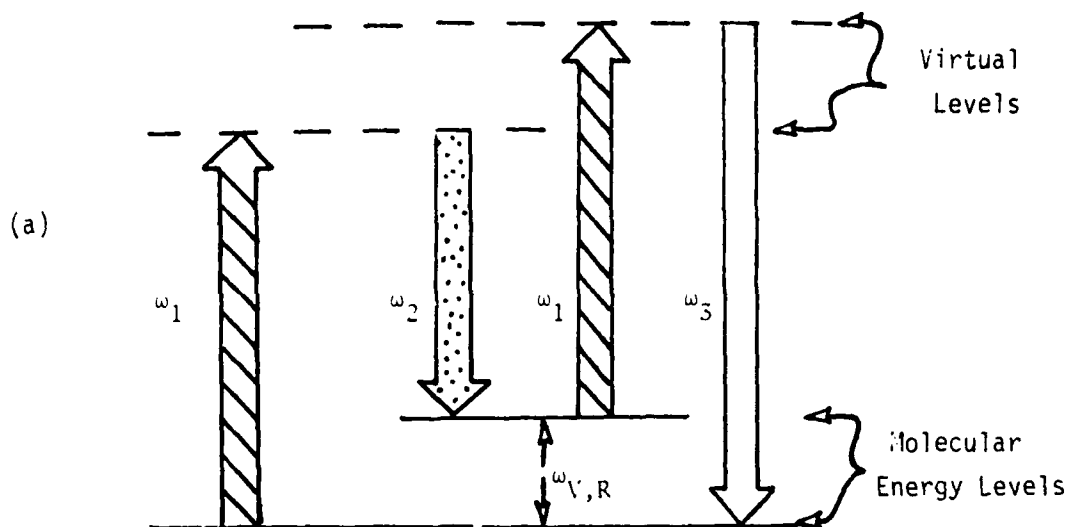


Figure 1. (a) Representative energy level diagram where  $\omega_1$  and  $\omega_2$  are laser frequencies,  $\omega_3$  is the CARS signal produced and  $\omega_{v,r}$  is an allowed molecular vibrational or rotational energy spacing. (b) Representative frequency spectrum when using broadband laser radiation centered at  $\omega_2$ .

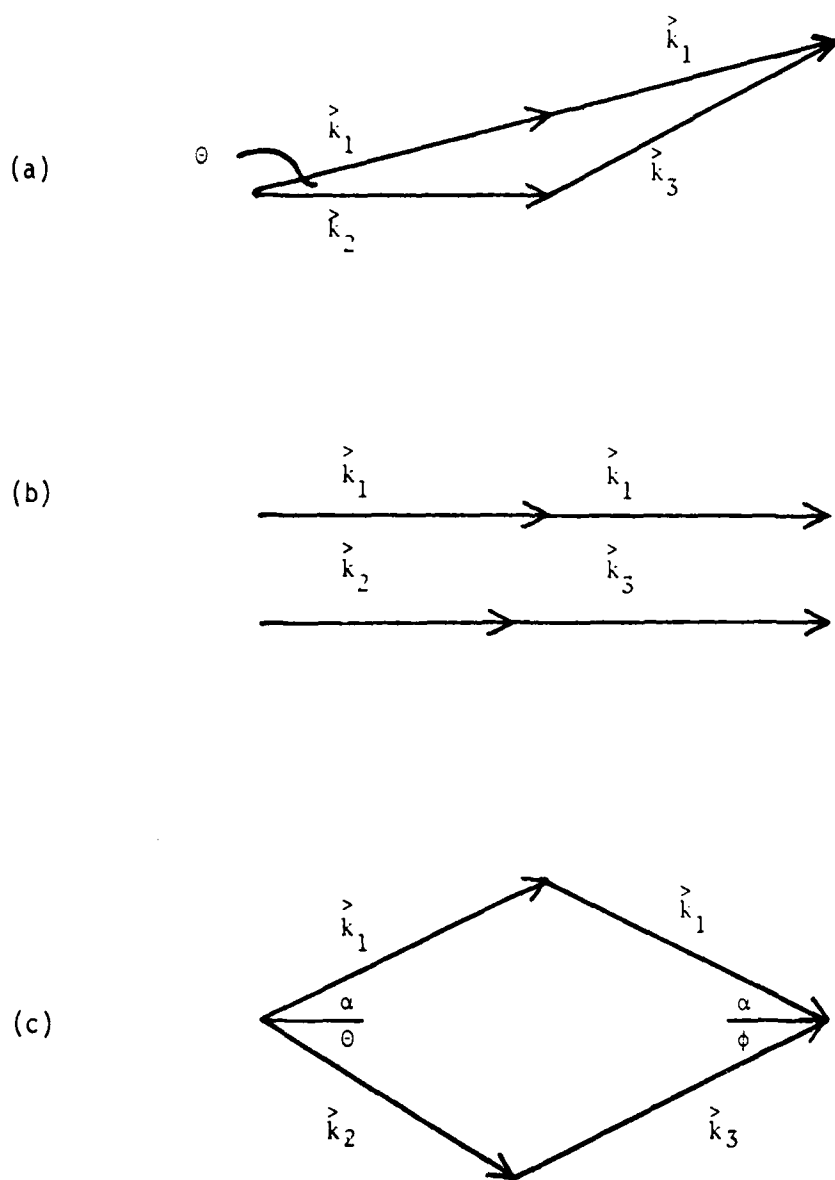


Figure 2. Three of the most common geometries used for phase matching ( $\Delta \vec{k} = 0$ ) of the laser beams (a) crossed beams, (b) collinear beams and (c) BOXCARS, where the  $\omega_1$  laser beam is split into two components. This geometry can result in much better spatial resolution than (a) or (b).

configurations were used for the experimental results presented in this paper. A variation of BOXCARS called folded BOXCARS has also recently been used in studies of rotational CARS<sup>25</sup>. In this case the  $k_1$  and  $k_2$  vectors are not in the same plane and thus the  $\omega_3$  CARS beam appears in a place well separated from both  $\omega_1$  beams. This configuration was not employed in this investigation.

The polarization of the incident laser beams must also be oriented in certain directions to produce substantial CARS signals. Generally the polarizations of the  $\omega_1$  and  $\omega_2$  beams are aligned to produce the most intense CARS signals<sup>26</sup>. In special circumstances, such as suppression of the non-resonant susceptibility which can mix with the resonant contribution,<sup>17,19,27</sup> other polarization techniques are used but they will not be discussed here. For this work only the configuration where all laser polarizations are aligned in the same plane was used.

### III. EXPERIMENTAL

A schematic diagram of the CARS experimental apparatus is shown on Fig. 3. A commercial Nd:YAG Quantel Model 481C laser was used as the excitation source. This laser produces approximately 500 mJ per pulse of energy at 532 nm when operated at 10 Hz. It has a pulse length of about 15 ns and a linewidth of  $\sim 0.2 \text{ cm}^{-1}$ . The  $\omega_1$  beam, vertically polarized,<sup>28</sup> is routed by dichroic mirrors M1-M6 which have maximum reflectance at 532 nm and maximum transmission at 1064 nm thus allowing separation of the fundamental from the experiment. The fundamental is terminated with a beam block (not shown). BS1 is a 50/50 beamsplitter which splits off half of the  $\omega_1$  beam for pumping the dye laser. A Quantel Model TDL III dye laser was used to supply the light at frequency  $\omega_2$ .<sup>28</sup> This dye laser, as obtained from the manufacturer, operates at narrow linewidths which are obtained with a tuning mechanism consisting of a grazing angle of incidence grating and return mirror. To create broadband radiation from this dye laser the optical path from the grating and return mirror was blocked. In addition a broad band high reflectance mirror was placed just past the grating which returned some of the zeroth order grating light back to the oscillator dye cell. Adjustment of this mirror allowed the dye laser to lase essentially over the complete dye gain curve.

<sup>25</sup>J.A. Shirley, R.J. Hall, and A.C. Eckbreth, "Folded BOXCARS for Rotational Raman Studies," *Optics Letters*, Vol. 5, 380-382, 1980.

<sup>26</sup>M.D. Levenson and N. Bloembergen, "Dispersion of the Nonlinear Optical Susceptibility Tensor in Centrosymmetric Media," *Phys. Rev. B* Vol. 10, pp. 4447-4463, 1974.

<sup>27</sup>R.L. Farrow, R.E. Mitchell, L.A. Rahn, and P.L. Mattern, "Crossed-Beam Background-Free CARS Measurements in a Methane Diffusion Flame," AIAA-81-0182, 1981.

<sup>28</sup>Polarization of  $\omega_1$  can be made either horizontal or vertical by orientation of the frequency doubling crystal. To obtain essentially one polarization in the dye laser a Glan-air prism can be inserted in the oscillator cavity and thus the polarization can be changed by orientation of this prism.

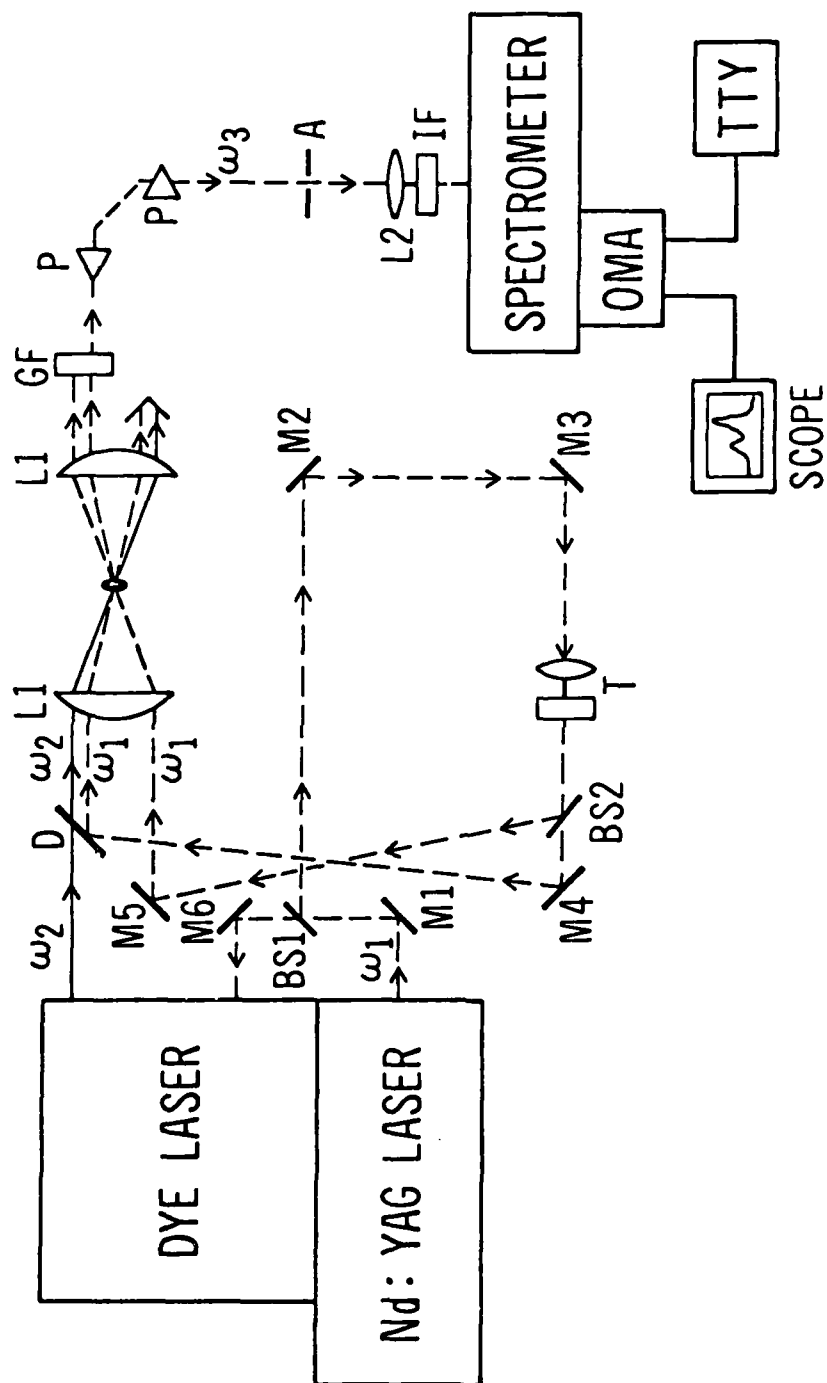


Figure 3. A Diagram of the Experimental Apparatus.

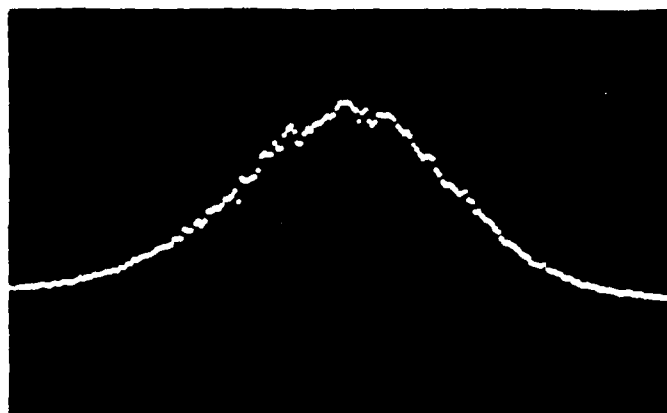
Since this dye laser consists of one oscillator and three amplifiers a substantial optical delay line (4m) was required to temporally match the  $\omega_2$  radiation with that of  $\omega_1$  at the point of intersection of the beams. This optical delay line is represented in Fig. 3 from BS1 through M4. BS2 is another 50/50 beamsplitter which splits the  $\omega_1$  beam for the BOXCARS configuration. D is a dichroic mirror which reflects the  $\omega_1$  (532 nm) radiation and allows most of the  $\omega_2$  radiation to pass through. When producing a CARS signal for  $N_2$ ,  $\omega_2$  is centered at ~606 nm, 680 nm for  $H_2$ , and D allows greater than 70% transmission at these wavelengths. This transmission is highly dependent on the angle of incidence of the  $\omega_2$  radiation.

L1 and L2 provide the focussing, intersection and recollimation of the  $\omega_1$  and  $\omega_2$  beams. These lenses are 7.6 cm in diameter with a 30.5 cm focal length. At the intersection point it is desirable for the beams to waist and have similar diameters for efficient CARS signal generation. A telescope, T, was used for this purpose. Several stages of filtering are used to eliminate the  $\omega_1$  and  $\omega_2$  radiation from  $\omega_3$ . Glass absorption filters, GF, are used to absorb > 95% of the  $\omega_1$  and  $\omega_2$  radiation while passing ~50% of the  $\omega_3$  signal. Spatial separation of the remaining beams is accomplished by using two extra dense flint prisms, P. Later these two prisms were replaced by a rutile prism. The  $\omega_3$  beam then passes through an aperture A and is focussed into a 0.25 m spectrometer by L2 which has a 7.6 cm focal length. An interference filter, IF, is placed just in front of the spectrometer entrance to remove any stray  $\omega_1$  and  $\omega_2$  light. This filter blocks ~99 % of 532 nm radiation and passes ~85% of the  $\omega_3$  radiation. The spectrometer was positioned such that the entrance slits, 100  $\mu$ , were horizontal, that is, parallel to the plane of the laser beams. This arrangement eliminated a potential problem of not capturing all of the requisite spectrum since the spatial dispersion, from the prisms, disperses the light along the long dimension of the entrance slit (1 cm). A silicon intensified target vidicon tube, PAR Corp. optical multichannel analyzer (OMAI) Model 1205D, was used to detect the dispersed light of the spectrometer. Either of two gratings, 1180 or 2360 grooves/mm, can be switch selected. Using a grating of 1180 grooves/mm one finds approximately 40 nm of radiation can be observed at one time. This radiation is dispersed on 500 memory channels which gives a resolution of 0.08 nm per channel.

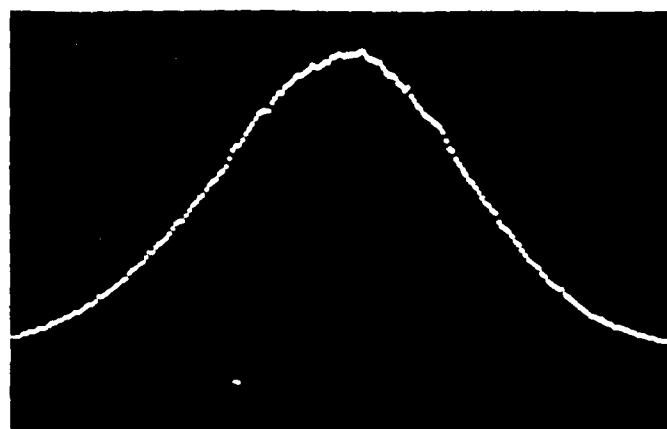
Two memories of the OMA allow summation of scans to obtain data from one or more laser pulses and also noise subtraction. In practice the CARS signal and background noise are recorded in memory A, then the CARS signal is removed by blocking the  $\omega_2$  beam and this background noise signal is recorded in memory B. Subtraction of B from A results in a relatively noise free CARS signal. The contents of the OMA can be observed on a display scope or digitally outputted on a teletype and paper tape.

To generate  $N_2$  CARS signals the dye laser was centered at  $\omega_2$  ~606 nm using one part Rhodamine 640 to four parts Rhodamine 610 by volume in a methanol solvent at concentrations of  $\sim 10^{-4}$  moles/liter. For  $H_2$  CARS signals,  $\omega_2$  is centered at 680 nm. Oxazine 720 dye in methanol at concentrations of  $\sim 5 \times 10^{-3}$  moles/liter produced the desired wavelengths. The full width half maximum (FWHM) bandwidth for the dye mixtures was about 5 nm. A photograph of the output from the OMA display scope showing the broadband output is shown in Figs. 4a and b. An easy method of determining where the dye laser is centered can be accomplished by operating the TDL III

(a)



(b)



(c)

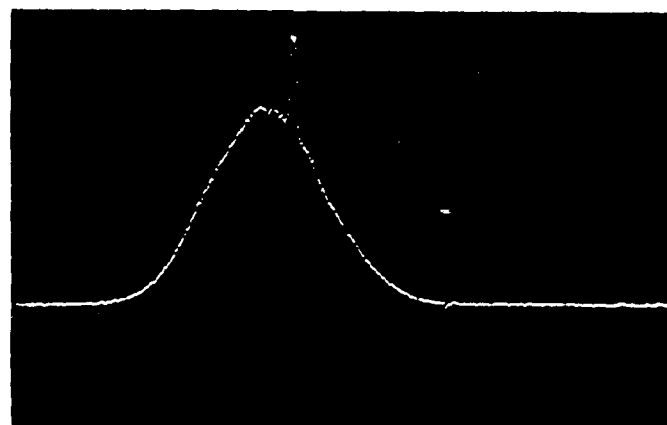


Figure 4. Frequency spectra of the dye laser. (a) Broadband dye laser output from a single laser shot. The peak is centered at 631 nm with a FWHM of about  $110 \text{ cm}^{-1}$ . (b) Same conditions as in (a) except 16 laser shots were accumulated. (c) The dye laser was operated such that a broadband and narrowband output was produced simultaneously for the purpose of checking the center frequency. Here the narrowband output is 633 nm as indicated by the calibrated wavelength dial of the dye laser.



dye laser in the broad and narrow band modes simultaneously (that is, allow the dye oscillator light from the grating to return to the oscillator cell as well as the light returned from the broad band reflector mirror). The dye laser output for this configuration is shown in Fig. 4c.

A porous plug burner made from sintered copper shown in Fig. 5 was used to support the various flames used in this study. The sintered plug diameter is 2.8 cm and is edge cooled by water flowing through a copper coil soldered to the burner case. All of these studies were conducted at atmospheric pressure.

The spatial resolution of the CARS experiment in the vertical direction (a plane parallel to the flow of burner gases) is given to a good approximation by the product of the focal length of the focussing lens (30.5 cm) and the divergence of the laser beam  $<0.5 \times 10^{-3}$  radians. This gives a vertical resolution of about 150  $\mu$ . The resolution in the horizontal direction (a plane perpendicular to the flow of burner gases) varies depending on whether collinear or BOXCARS geometry is used. Furthermore for BOXCARS geometry the resolution depends on the crossing angle of the laser beams. For the collinear case, assuming no room air contributions arise, the horizontal resolution is about 6 cm. The BOXCARS configuration is capable of producing much better horizontal resolution. Although the resolution was not investigated in detail here, Eckbreth<sup>24</sup> measured a resolution of 4 mm (the spatial extent where the CARS signal grew from 10% to 90%) for a geometrical arrangement similar to our investigations. Higher spatial resolution was not attempted or necessary for the results reported here; however, spatial resolutions covering volumes of less than 1 mm<sup>3</sup> should be easily attainable. If sufficient data can be obtained in a single laser pulse the temporal resolution is  $10^{-8}$  seconds.

#### IV. ANALYSIS OF CARS SPECTRA

The fitting routine used to reduce the experimental CARS spectra is a modification of a least squares program which has been previously used to extract spectroscopic constants from measured transition frequencies.<sup>29</sup> The program uses the equation

$$I_o(\omega_1^o, \omega_2^o, \nu_o) = \int d\omega |\chi(\omega)|^2 \exp \left[ -\frac{AB}{A+B} (\omega - \omega_o)^2 \right] \int d\omega_3 S(\nu_o, \omega_3) \exp \left[ -A(\omega_3 - \omega - \omega_1^o)^2 \right] \quad (1)$$

to fit the N<sub>2</sub> and H<sub>2</sub> CARS spectra obtained from laboratory flames. In Eq. 1, I<sub>o</sub> is the observed CARS signal intensity,  $\omega_1^o$  and  $\omega_2^o$  are the center frequencies of the pump and Stokes laser profiles respectively and  $\omega_o = \omega_1^o - \omega_2^o$ ,  $\nu_o$  is

<sup>29</sup>A.J. Kotlar, R.W. Field, J.I. Steinfeld and J.A. Coxon, "Analysis of Perturbations in the A<sup>2</sup> $\pi$  - X<sup>2</sup> $\Sigma^+$  Red System of CN," Journal of Molecular Spectroscopy, Vol. 80, p. 86-108, 1980.

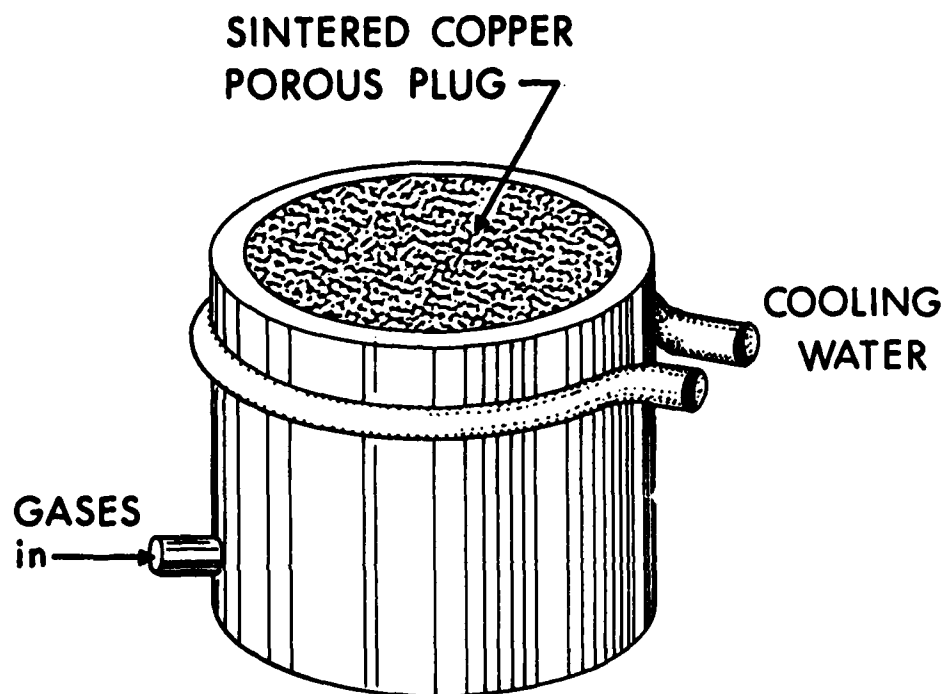


Figure 5. An edge cooled porous plug burner used to support the flames investigated in this report.

the center frequency of the detector slit function  $S$ , and  $A$  and  $B$  are related to the widths of the  $\omega_1$  and  $\omega_2$  laser profiles, which are assumed to be Gaussian.  $A = 4 \ln 2 / (\Delta\omega_1)^2$  and  $B = 4 \ln 2 / (\Delta\omega_2)^2$  where  $\Delta\omega_1$  and  $\Delta\omega_2$  are the laser widths (FWHM). A more detailed description of the CARS equation and commonly used slit width functions can be found in Ref. 30. The computer program performs a multiparameter fitting to obtain a best fit to the data. The program also calculates the variance/covariance matrix from which the standard deviation, as statistically determined from the fit, is evaluated.

For nitrogen spectra, typically six parameters were varied in the fit. These included the temperature, a scaling factor for the spectrum which corresponds to the number density, a baseline, the slit width, the resolution of the detector ( $\text{\AA}/\text{channel}$  for the OMA) and a reference wavelength to tie the spectrum to an absolute scale.

The spectrum for hydrogen is much better resolved than for the nitrogen case; thus the pressure broadened rotational linewidth was also adjusted in the fit. For some of the hydrogen experiments ( $\text{CH}_4$  diffusion flame and the rich  $\text{CH}_4/\text{N}_2\text{O}$  premixed flame) a large background is present due to collinear phase matching and low relative concentrations of  $\text{H}_2$ . As will be discussed later, this background feature is determined by the non-resonant susceptibilities and the position and shape of the dye laser profile. To properly account for this spectral feature in the computer fit four additional parameters are required. These parameters are the non-resonant susceptibility of the flame gases, the non-resonant susceptibility of room air, the center frequency and width of the dye laser profile. These parameters can be experimentally determined or found in the literature, however they were not constrained to one value, but rather allowed to be changed by the least squares fitting procedure.

The spectroscopic constants for nitrogen were taken from Huber and Herzberg<sup>31</sup>. The spectrum was calculated for all O, Q and S transitions for vibrational levels  $v'' = 0$  to 4 and rotational levels  $J'' = 0$  to 60. For the region of the spectrum corresponding to the range of the experimental data, transitions from 2269 to 2350  $\text{cm}^{-1}$  must be included in the fitting program. This gives 153 transitions from (3,2) Q(13) to (1,0) S(1). The spectroscopic constants for hydrogen were also taken from Ref. 31, where the higher order constants are from Fink et al.<sup>32</sup> Reference is also made to more recent values

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<sup>30</sup>A.J. Kotlar and J.A. Vanderhoff, "A Model for the Interpretation of CARS Experimental Profiles," accepted for publication in *Applied Spectroscopy*. "Interpretation and Least Squares Fitting of CARS Experimental Profiles," BRL report in preparation.

<sup>31</sup>K.P. Huber and G. Herzberg, "Molecular Spectra and Molecular Structure IV. Constants of Diatomic Molecules," Van Nostrand Reinhold Company, New York 1979.

<sup>32</sup>U. Fink, T.A. Wiggins and D.H. Rank, "Frequency and Intensity Measurements on the Quadrupole Spectrum of Molecular Hydrogen," *J. Mol. Spectrosc.* Vol. 18, p. 384, 1965.

by the same workers<sup>33</sup>. We found that the earlier values<sup>32</sup>, that is those actually listed in Huber and Herzberg, gave transition frequencies which better agreed with those we observed in our experiments, and these were used to generate the table of transitions used in the program. For the hydrogen spectrum corresponding to the range of experimental data 3940 to 4200  $\text{cm}^{-1}$ , it was only necessary to include vibrational levels  $v'' = 0$  and 1, and rotational levels  $J'' = 0$  to 8. Within this range only the (1,0) Q branch is present giving 9 transitions from Q(0) to Q(8).

## V. RESULTS

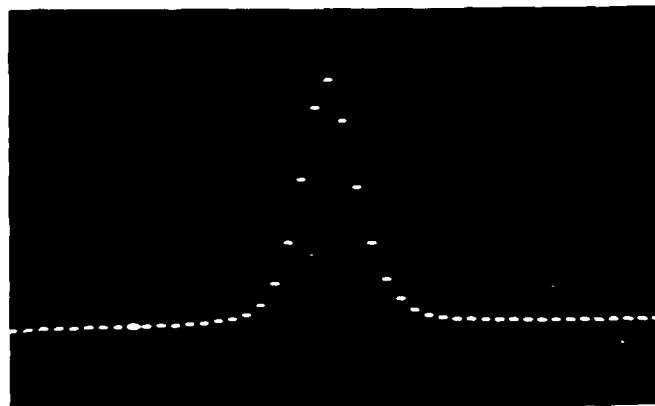
Initially collinear CARS was employed to survey qualitatively the shape and characteristics of various CARS spectra. No temperature fitting was attempted at this point. Fig. 6a shows a room air  $\text{N}_2$  CARS signal which has a relatively symmetric shape with a FWHM resolution of about 6  $\text{cm}^{-1}$ . Figs. 6b and 6c show the  $\text{N}_2$  CARS signal obtained in a slightly rich  $\text{CH}_4/\text{N}_2\text{O}$  flame without and with shields, respectively. The shields were hollow tubes placed between (see Fig. 3) the focussing lens and the burner and between the burner and the recollimating lens. Helium gas was flowed through these tubes to displace the room air. The effect of shielding is readily apparent from the differences observed in Figs. 6b and 6c. Fig. 6c has a distinct second peak, a hot band denoting a higher  $\text{N}_2$  temperature. Since these shields proved troublesome, it was necessary to go to the BOXCARS configuration when room air contributions could be present (i.e. the study of  $\text{N}_2$  or  $\text{O}_2$  molecules).

The signal strength for BOXCARS is less than for collinear geometry due to a smaller interaction length; thus we chose to first look at a near stoichiometric  $\text{CH}_4/\text{air}$  flame where the  $\text{N}_2$  density is larger than for a  $\text{CH}_4/\text{N}_2\text{O}$  flame. Figures 7a,b show BOXCARS spectra for  $\text{N}_2$  taken at two positions perpendicular to the plane of the burner surface. The temperature values with standard deviations are listed in Table 1. A crossing angle ( $2\alpha$ ) of  $6^\circ$  was used. The 2360 groove/mm grating gave a FWHM resolution of 3.8  $\text{cm}^{-1}$ . This resolution is sensitive to the focussing of the lens L2 and is determined from the CARS computer fitting program by best fits to the data especially on room air where the temperature is precisely known. The resulting temperature profile is suggestive of several phenomena associated with the flame. First, at the 1 mm position the temperature is much lower than the other (higher) positions indicating that we are sampling somewhere in the reaction zone, a place prior to complete combustion. Second, the highest temperature measured is 1790K whereas the adiabatic flame temperature for a stoichiometric  $\text{CH}_4/\text{air}$  flame at atmospheric pressure is calculated from the NASA-Lewis thermochemical equilibrium code<sup>34</sup> to be 2226K. A substantial amount of heat is thus being extracted from the flame by the burner head<sup>35</sup>. Third, the burnt gas region of

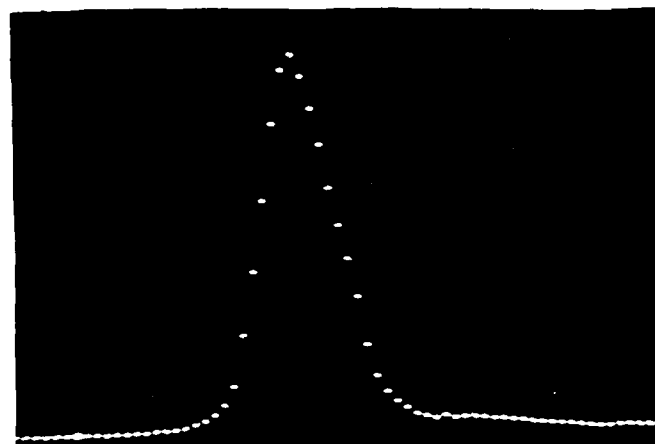
<sup>33</sup>J.V. Flotz, D.H. Rank and T.A. Wiggins, "Determinations of Some Hydrogen Molecular Constants", *J. Mol. Spectrosc.* Vol. 21, p. 203, 1966.

<sup>34</sup>R.A. Svehla and E.J. McBride, "Fortran IV Computer Program for Calculation of Thermodynamic and Transport Properties of Complex Chemical Systems", NASA TN D-7056 (1973).

(a)



(b)

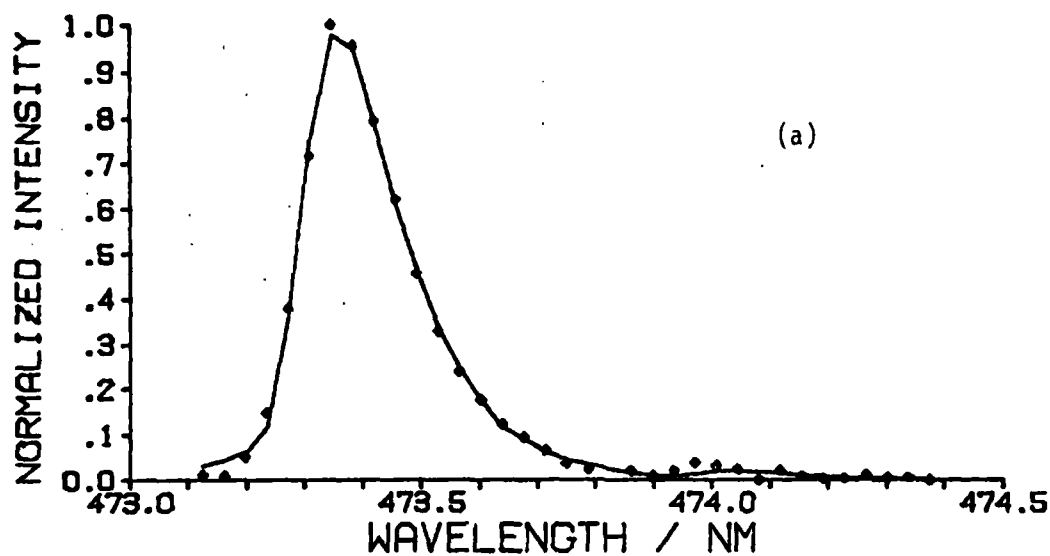


(c)



Figure 6. Photographs of the  $N_2$  CARS signal obtained in room air and a flame environment using collinear geometry. (a) room air, (b) a rich  $CH_4/O_2$  flame without shielding from air contributions, and (c) the same flame as in (b) except shields were used to eliminate room air contributions.

$T=1197(45)$



$T=1790(37)$

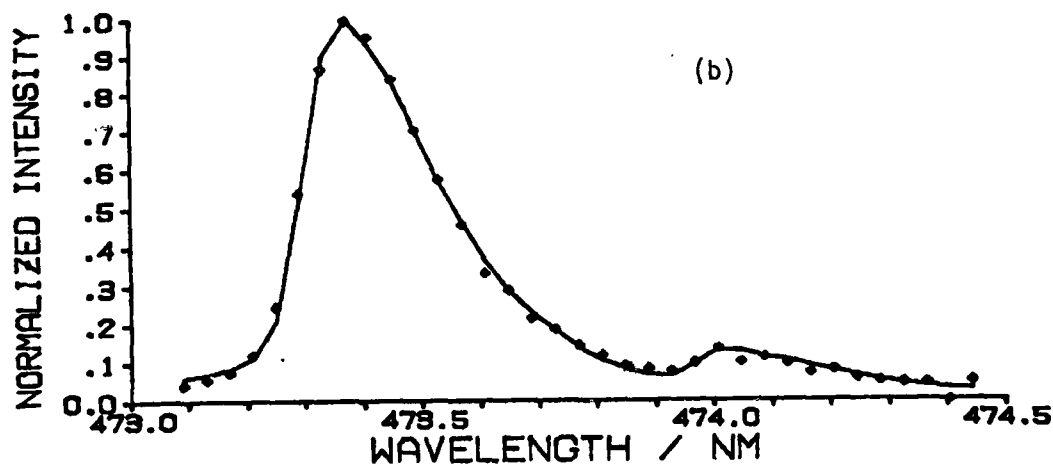


Figure 7a,b.  $N_2$  CARS spectra obtained in a near stoichiometric  $CH_4$ /air flame using BOXCARS geometry and a 2360 groove/mm grating. The experimental data are shown as points and the solid line is the least squares computer fit. The temperature resulting from the fit and the standard deviation are given on each spectrum. This holds for all subsequent CARS spectra.

Height (mm)	Temperature (K)
1	1197±45
4	1790±37
9	1645±51
14	1553±39
19	1687±44

Table 1. Vertical temperature profile values obtained in a near stoichiometric  $\text{CH}_4$ /air flame using BOXCARS geometry with a crossing angle of  $6^\circ$  ( $2\alpha$ ). A 2360 groove/mm grating was used giving a resolution of about  $4 \text{ cm}^{-1}$  (FWHM). The least squares computer fits which are tabulated in the three tables have the resolution as a variable parameter to be determined by the best fit thus the resolutions quoted here are average values which may vary as much as  $\pm 3 \text{ cm}^{-1}$  from run to run. The number of laser shots for each temperature determination varied from 5 to 50 shots.

the flame (4 mm through 19 mm) does not show either a constant temperature or a slow fall off but instead a decrease followed by an increase in temperature. This behavior is indicative of a second flame zone which could result if the flame is slightly rich and a secondary zone forms from combustion with entrained air.

The next set of CARS measurements were made on a near stoichiometric  $\text{CH}_4/\text{N}_2\text{O}$  flame using BOXCARS geometry with a crossing angle of  $3.9^\circ$ . A vertical and radial temperature profile is illustrated on Figures 8a and 8b with values listed in Table 2. Each data point was a sum of 300 laser pulses and representative CARS spectra for two of these data points are shown on Fig. 9a and 9b. Here a 1180 groove/mm grating was used resulting in the signal being displayed on half as many OMA channels as the data of Figs. 7 and 11. With the exception of the positions higher than 10 mm and at the edges of the burner the temperatures agree within the constraints of the standard deviation (see Table 2). Above 10 mm the flame is cooled by heat loss to the surrounding air. At the edges of the burner where steep gradients are present, it is not surprising to find differences in the temperature results. Another vertical and radial temperature profile in a near stoichiometric  $\text{CH}_4/\text{N}_2\text{O}$  flame was obtained using the 2360 groove/mm grating. The profiles are shown in Figs. 10a and 10b, and the values are given in Table 3. Representative CARS spectra for two of these data points are shown in Figs. 11a and 11b.

It is interesting to compare the results of Figs. 8 and 10. These data were obtained on different days; hence the equipment and flame were subject to the routine tasks of turn-off-turn-on and alignment between data sets. The flame temperatures measured from 0 - 10 mm in the vertical direction and close to the centerline radially exhibit excellent agreement between data sets. Above 10 mm and close to the edge radially the temperatures differ substantially. This effect is most likely caused by the chimney which captures the burnt gases of the flame. The closer the chimney is placed to the burner the more the flame contracts to the centerline of the burner as a function of vertical height above the burner surface. Although the airflow up the chimney provides for stabilization of the flame against fluctuating room air currents it also changes the geometry of the flame near the edges of the burner and causes greater cooling high above the burner surface. The data for Fig. 10b were taken with the chimney closer to the burner resulting in the steeper gradients observed.

By comparing the standard deviations of the temperatures shown in Tables 2 and 3 one observes that the deviation is larger for the lower resolution data which is expected. The resolution obtained with the 1180 groove/mm grating ( $\sim 12 \text{ cm}^{-1}$ ) is about the lowest resolution that can be used and still obtain reasonable precision at flame temperature ( $\sim 2000\text{K}$ ) for  $\text{N}_2$  CARS spectra. At lower temperatures higher resolution is necessary since smaller contributions from the  $\text{N}_2$  hot band occur and thus the temperature must be determined predominantly from the shape of the single peak.

<sup>35</sup> This effect has been observed in our laboratory whenever using porous plug burners in spontaneous Raman and fluorescence studies. See for example W.R. Anderson, L.J. Decker, and A.J. Kotlar, "Temperature Profile of a Stoichiometric  $\text{CH}_4/\text{N}_2\text{O}$  Flame from Laser Excited Fluorescence Measurements on OH," accepted for publication in Combustion and Flame.



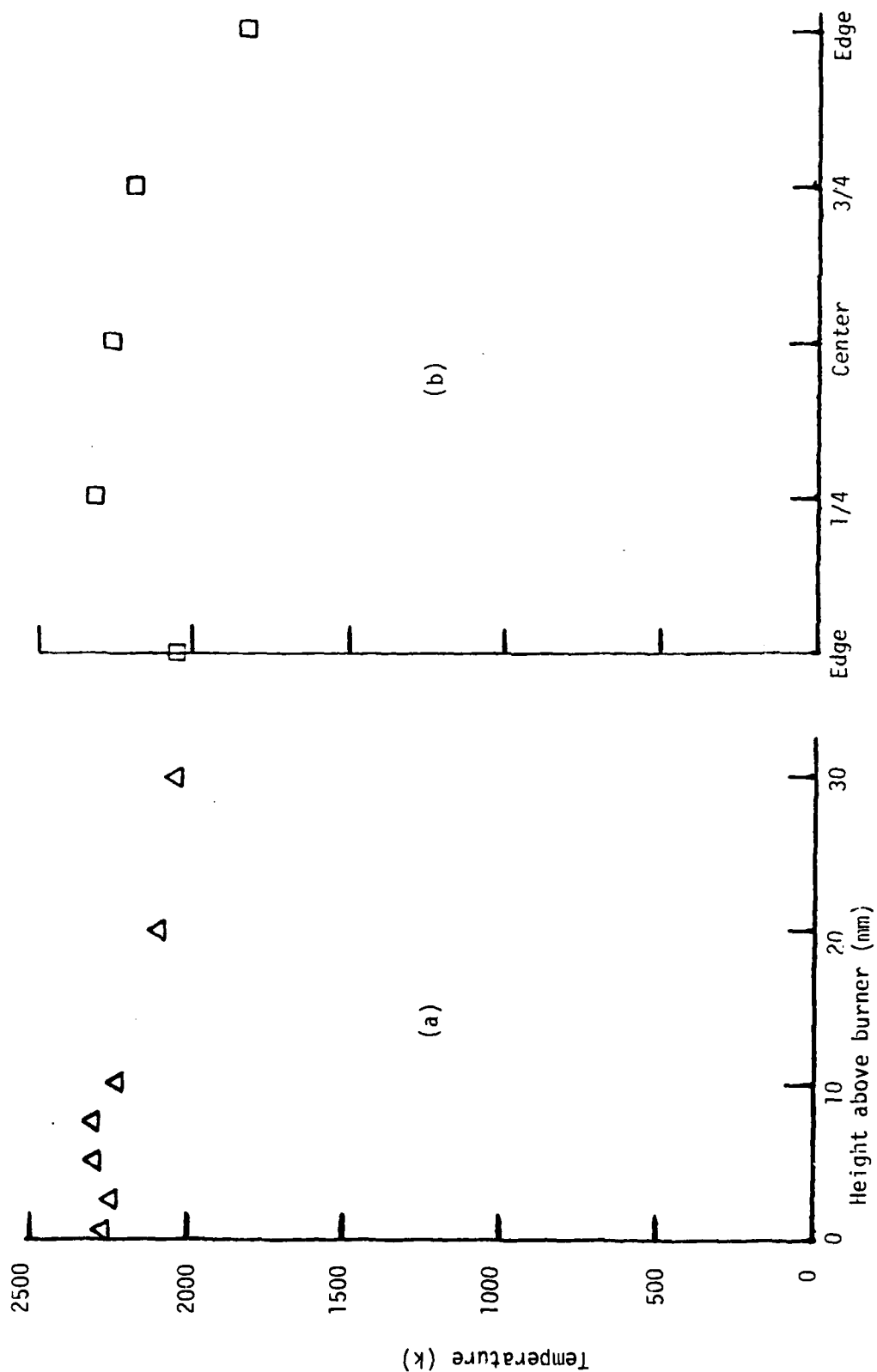
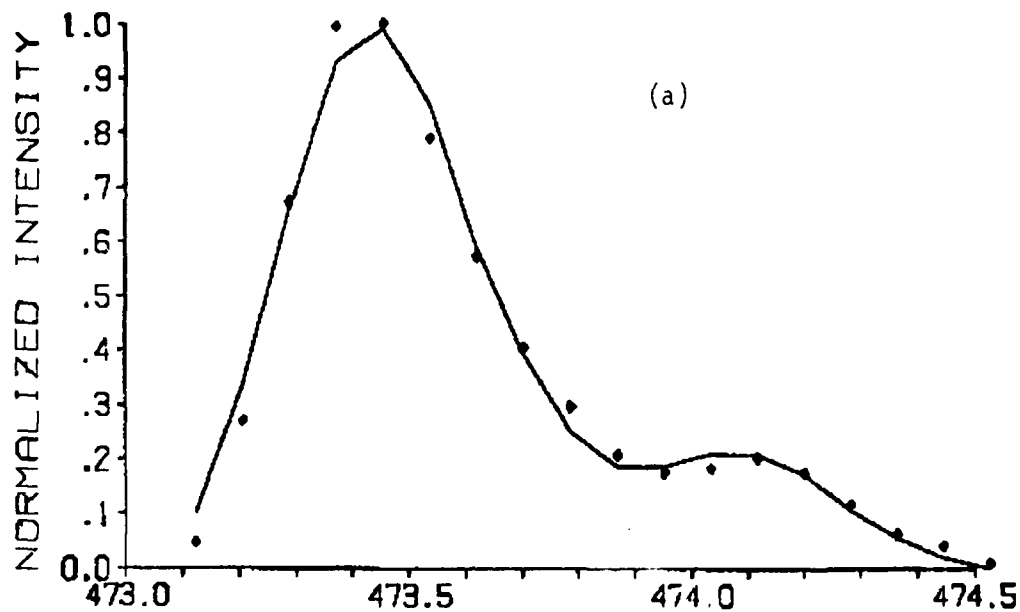


Figure 8.  $\text{N}_2$  CARS temperature profiles obtained in a near-stoichiometric  $\text{CH}_4/\text{N}_2\text{O}$  flame; (a) vertical profile (b) radial profile.

Height (mm)	Radial Position	Temperature (K)
0.5	centerline	2271±73
2.5	centerline	2249±38
5.0	centerline	2304±50
7.5	centerline	2309±37
10.0	centerline	2234±40
20.0	centerline	2101±4
30.0	centerline	2068±19
5.0	edge	2083±45
5.0	1/4	2333±90
5.0	centerline	2278±77
5.0	3/4	2209±39
5.0	edge	1841±36

Table 2. Vertical and radial temperature profile values obtained in a near-stoichiometric  $\text{CH}_4/\text{N}_2\text{O}$  flame using BOXCARS geometry with a crossing angle of  $3.9^\circ$ . A 1180 groove/mm grating was used giving a resolution of  $12 \text{ cm}^{-1}$ . An accumulation of 300 laser shots per temperature point was used.

$T=2083(45) \text{ K}$



$T=2304(50) \text{ K}$

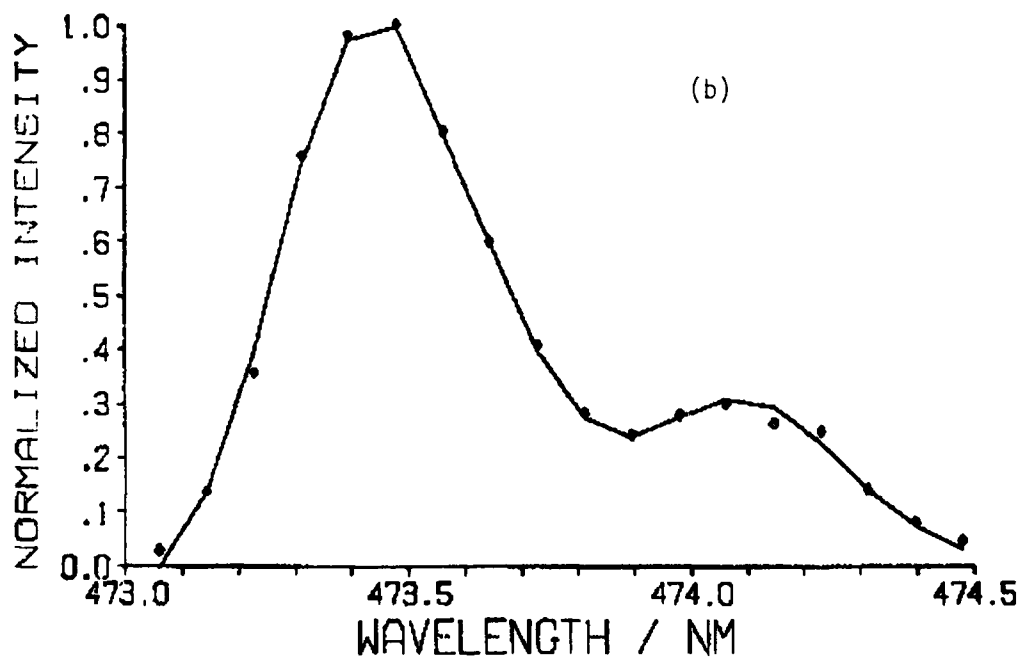


Figure 9a,b. Two representative N<sub>2</sub> CARS spectra for two of the data points of Fig. 3. BOXCARS geometry and a 1180 groove/mm grating was used.

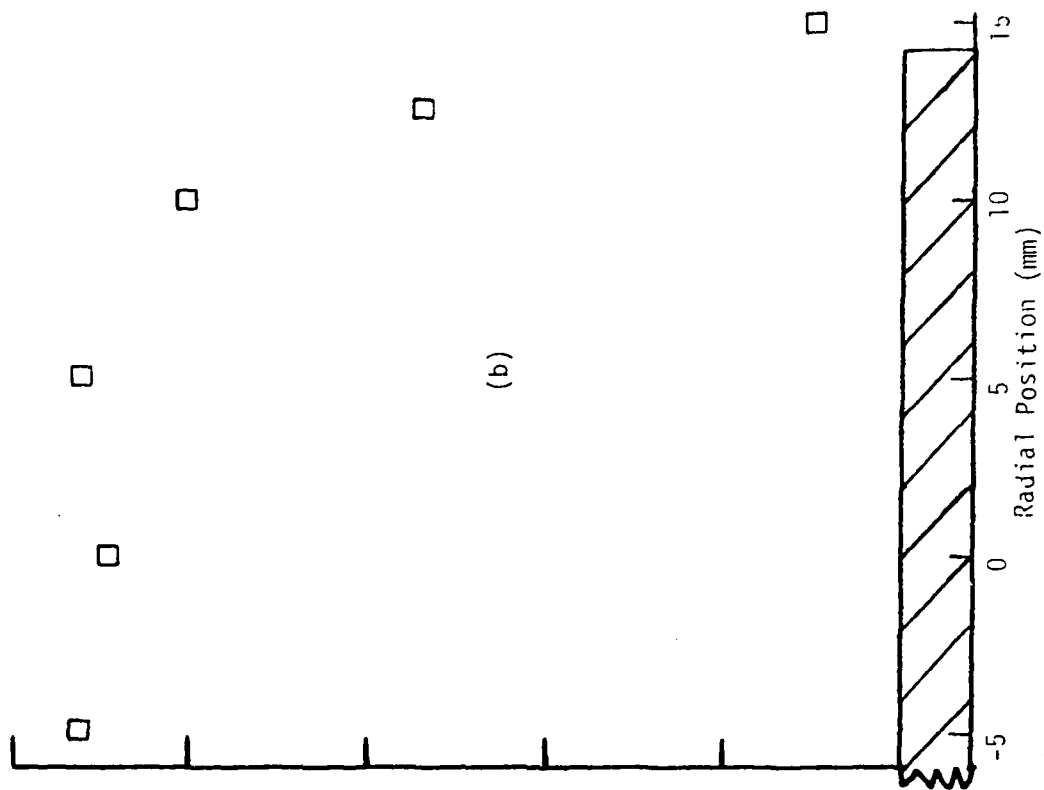
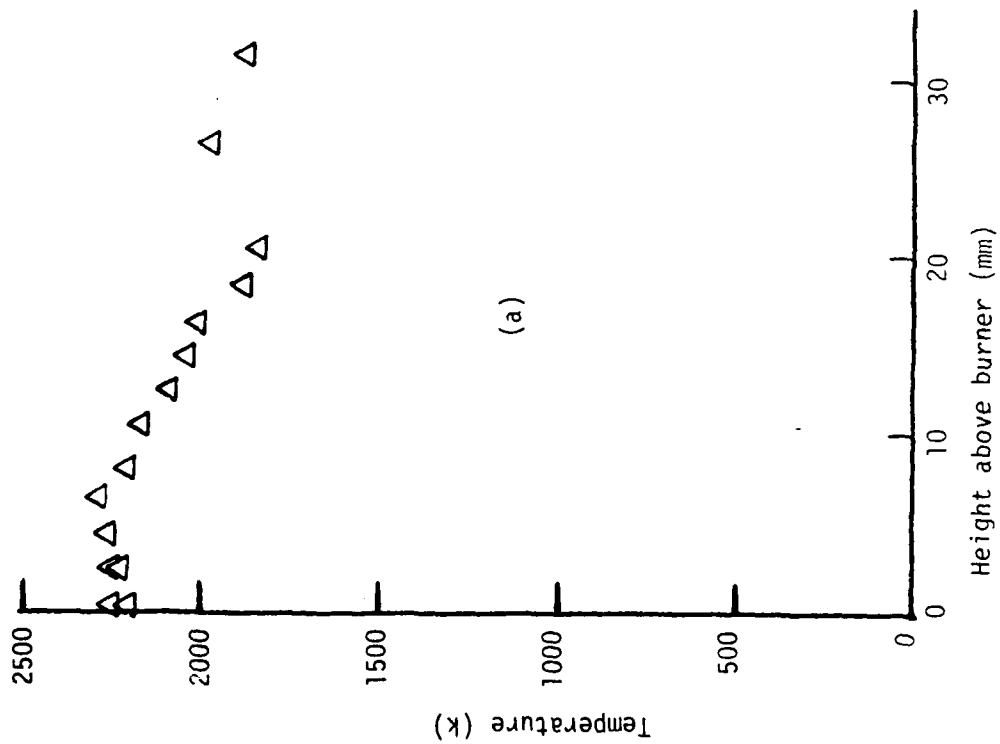


Figure 10. N<sub>2</sub> CARS temperature profiles obtained in a near-stoichiometric C<sub>2</sub>H<sub>4</sub>/H<sub>2</sub>O flame;  
(a) vertical profile (b) radial profile.

Height (mm)	Radial Position (mm)	Temperature (K)
0.5	centerline	2270±10
0.5	centerline	2278±30
0.5	centerline	2226±37
2.5	centerline	2234±35
2.5	centerline	2250±40
4.5	centerline	2271±38
6.5	centerline	2302±29
8.5	centerline	2208±28
10.5	centerline	2181±52
12.5	centerline	2096±23
14.5	centerline	2040±23
16.5	centerline	2016±24
18.5	centerline	1892±22
20.5	centerline	1855±54
26.5	centerline	1985±40
31.5	centerline	1880±26
2.5	-5	2320±45
2.5	+5	2315±40
2.5	+10	2017±24
2.5	12.5	1343±19
2.5	15	227±30

Table 3. Vertical and radial temperature profile values obtained in a near-stoichiometric  $\text{CH}_4/\text{N}_2\text{O}$  flame using BOXCARs geometry with a crossing angle of  $3.9^\circ$ . The 2360 groove/mm grating gave an overall resolution of  $4\text{ cm}^{-1}$ . An accumulation of 650 laser shots were used per temperature point.

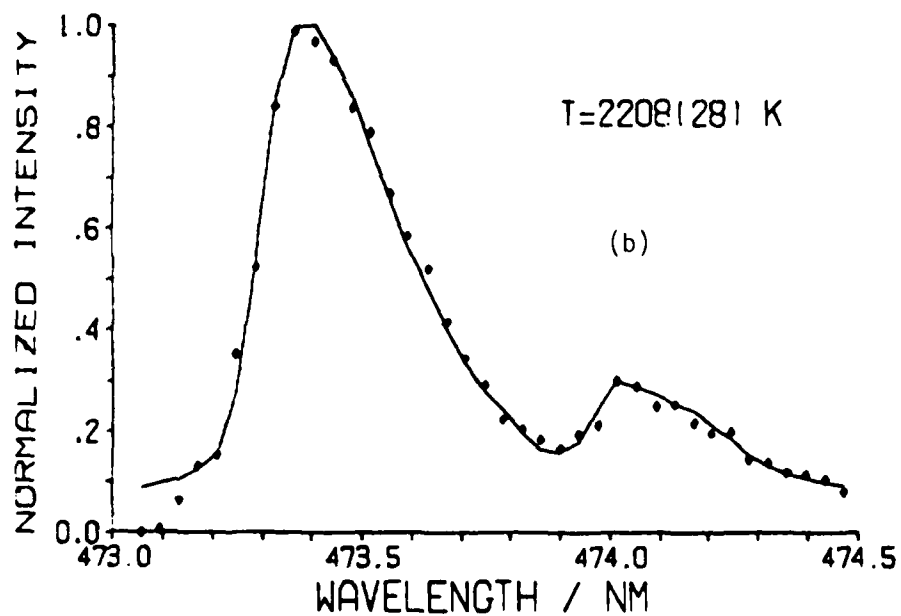
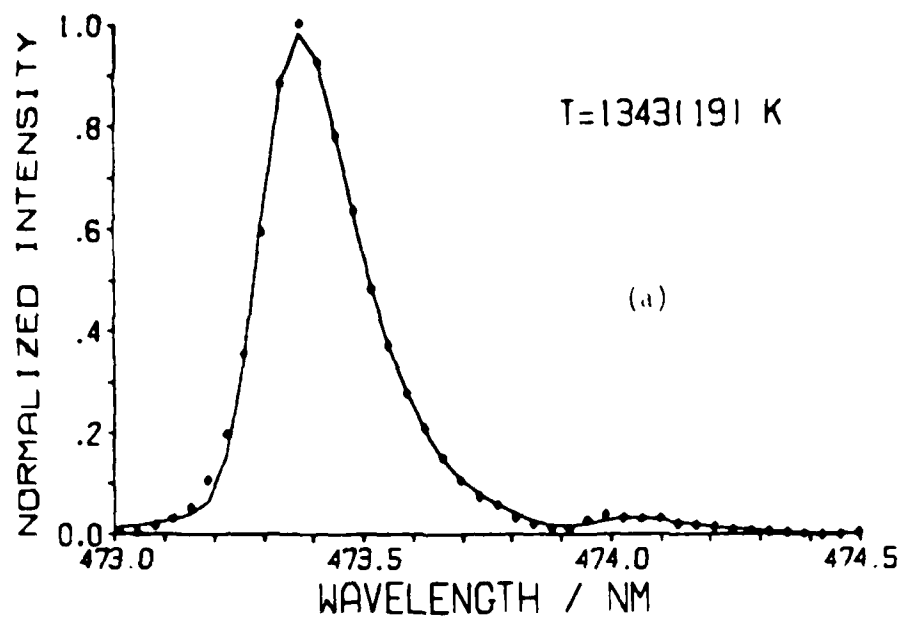


Figure 11a,b. Two representative  $\text{H}_2$  CARS spectra for two of the data points of Fig. 10. BOXCARS geometry was used and a 2360 groove/mm grating.

Temperature determinations were also made from CARS spectra of the  $H_2$  molecule in a flame environment.  $H_2$  has a well separated rotational spectrum due to its small moment of inertia ( $H_2$  has a rotational constant of  $60.8 \text{ cm}^{-1}$  while  $N_2$  is  $2.01 \text{ cm}^{-1}$ )<sup>36</sup> and thus the CARS spectrum of  $H_2$  shows well resolved rotational structure within the 0 to 1 vibrational Q-branch transition. Figs. 12a-d illustrate  $H_2$  CARS spectra obtained in (a) pure  $H_2$  at room temperature, (b) a  $H_2$  diffusion flame, (c) a  $CH_4$  diffusion flame and (d) a rich  $CH_4/N_2O$  premixed flame. In order to generate the spectrum of Fig. 12a pure  $H_2$  was flowed unignited through the porous plug burner and measurements were made 4 mm above the burner surface using BOXCARS geometry with a crossing angle of  $3.9^\circ$ . The resulting  $H_2$  CARS signal was easily observable with the eye and had to be attenuated by 99% before entering the detector. The generated spectrum is from a single shot. The largest peak at this temperature is  $Q(1)$ , i.e. the  $J = 1$  rotational line of the Q branch.  $Q(0)$  appears as a shoulder to the left of  $Q(1)$ , and  $Q(2)$  and  $Q(3)$  are also observable. The computer model code for  $H_2$  was constrained to fit this data with the room temperature value of 290K. In Fig. 12b the  $H_2$  flowing through the porous plug burner was ignited producing a  $H_2$ /air diffusion flame from which a temperature determination was made at a position 40 mm above surface<sup>37</sup>. This position appeared visually to be the hottest part of the diffusion flame. Here the CARS signal was substantially smaller and Fig. 12b represents 160 laser shots with no attenuation in the CARS signal beam. BOXCARS geometry was again used with the  $3.9^\circ$  crossing angle. At this hotter temperature the higher rotational lines  $Q(4)$  through  $Q(7)$  are observable. Higher rotational lines would also be observed if the dye laser had a broader bandwidth or shifted to higher wavelengths. A temperature of  $2406 \pm 58 \text{ K}$  was obtained from the least squares computer fit. This experimentally determined temperature is higher than the adiabatic flame temperature, 2382K, computed for a stoichiometric  $H_2$ /air flame from the NASA-Lewis code.<sup>34</sup> Since the investigation of  $H_2$  CARS signals in flame environments was of a survey nature we did not do any detailed investigation of this difference. Nonetheless, possible explanations for this difference are that the center frequency of the dye laser shifted from the calibrated value, or the linewidth and lineshape for  $H_2$  are different from those used in the computer program. In the computer program linewidths (Lorentzian) for  $N_2$  were fixed at  $0.1 \text{ cm}^{-1}$ , for  $H_2$  varied from 0.09 to  $0.15 \text{ cm}^{-1}$ . For  $H_2$  a Voigt profile was used for the lineshapes. At these temperatures ( $\sim 2000 \text{ K}$ )  $H_2$  falls into the Dicke narrowing regime and the chosen lineshape may not be strictly valid. A discussion of the variation of  $H_2$  linewidth with pressure and temperature can be found in Ref. 38.

<sup>36</sup>G. Herzberg, Spectra of Diatomic Molecules, (Van Nostrand Reinhold, New York, 1950).

<sup>37</sup>It was stated earlier that the burner head cooled the flame substantially. This holds true only for premixed flames where the flame is stabilized on the burner head. In the diffusion flames where room air is the oxidant, the reaction zone is cone shaped due to the way the air mixes with the fuel. Around the tip of the cone (40 mm) where the measurement was made there is negligible heat loss to the burner.

<sup>38</sup>J.A. Shirley, A.C. Eckbreth, and R.J. Hall, "Investigation of the Feasibility of CARS Measurements in Scramjet Combustion," UTRC Report R79-954390-8, 1979.

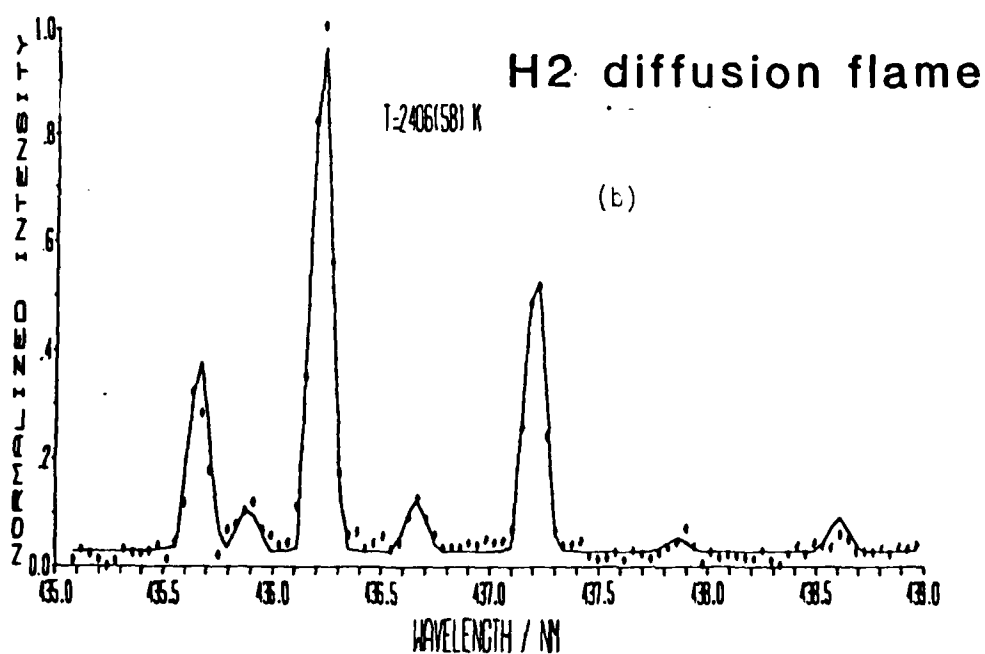
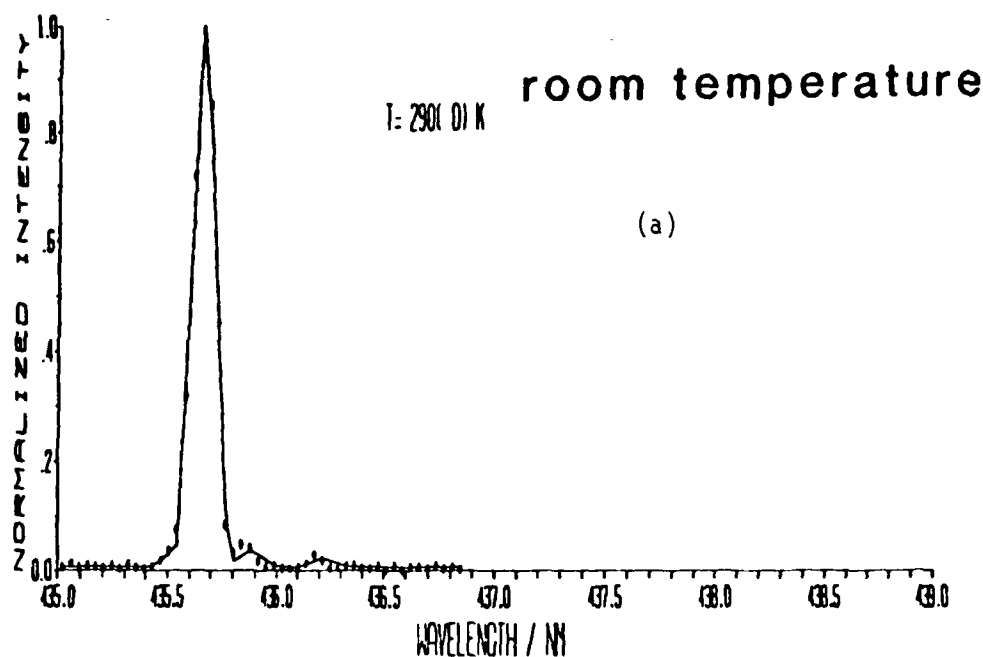


Figure 12. H<sub>2</sub> CARS spectra obtained in various environments using a 2360 groove/mm grating giving a resolution of about  $5 \text{ cm}^{-1}$ ; (a) pure H<sub>2</sub> at room temperature using BOXCAR geometry, (b) a H<sub>2</sub> diffusion flame using BOXCAR geometry.



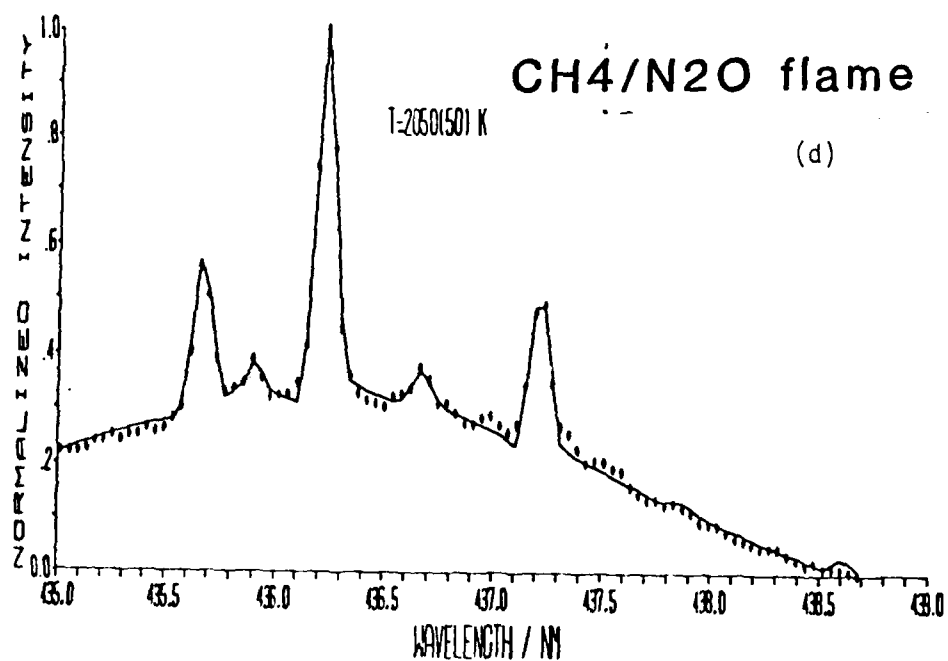
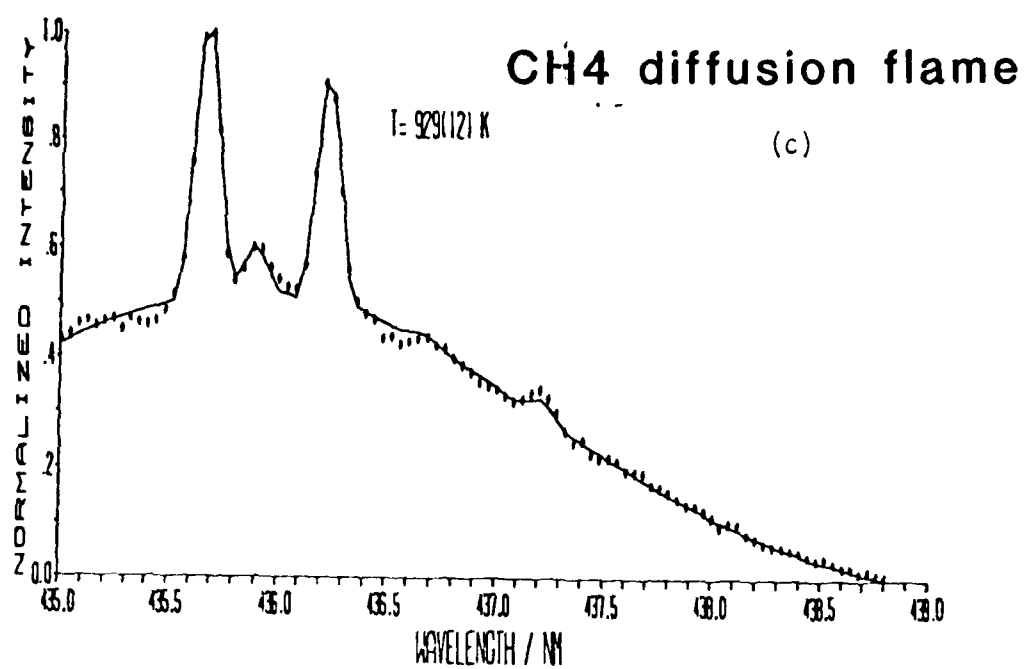


Figure 12 continued; (c) a CH<sub>4</sub> diffusion flame using collinear geometry and (d) a rich CH<sub>4</sub>/N<sub>2</sub>O flame using collinear geometry.

Figs. 12c and 12d show  $H_2$  CARS signals obtained in a relatively cool part of a  $CH_4$  diffusion flame and a rich  $CH_4/N_2O$  flame, respectively. In these cases much smaller quantities of  $H_2$  were present for CARS signal generation and collinear geometry was employed to obtain sufficient signal. The measurements were made 6 mm above the porous plug burner surface using 33 laser shots and an unattenuated CARS signal beam. The broad underlying background is the manifestation of the CARS signal generated by the non-resonant susceptibility. A useful feature of this broad curve is that it tracks the dye laser output which is now conveniently observed at the anti-Stokes frequency and thus can be used to gauge exactly where the dye laser is peaked as well as its overall shape for that experimental run. It can be seen that the  $H_2$  spectra in Figs. 12c and 12d are not perturbed to any extent by the non-resonant susceptibility other than sitting on a broad background. This observation can be explained as follows. In collinear geometry the laser beams are also overlapped in the room air surrounding the flame and this results in generation of CARS signals. However, the frequencies of the lasers have been set for resonance conditions for  $H_2$ , thus only non-resonance CARS signals appear from the room air ( $N_2$  and  $O_2$ ). This is the dominant contribution to the non-resonant signal, a place where there is no  $H_2$ , thus no mixing of resonant  $H_2$  CARS signals occur. If the dominant non-resonant contribution were occurring inside the flame where  $H_2$  is also present, then an obvious perturbation of the  $H_2$  spectrum would result. The computer fits for Figs. 12c and 12d include only a non-resonant susceptibility contribution outside the flame. An attempt to fit the experimental data of Fig. 12d with the non-resonant contribution occurring only inside the flame is shown in Fig. 13. Clearly this results in a poor fit and an unreasonably high temperature. The double asterisk for the standard deviation indicates that more than two digits are required.

## VI. CONCLUSIONS AND RECOMMENDATIONS

CARS spectroscopy has been used to determine the temperature and temperature profiles in several laboratory flames. CARS signals from both  $N_2$  and  $H_2$  have been used in these determinations. This technique is capable of high temporal ( $10^{-8}$ s) and spatial ( $1\text{mm}^3$ ) resolution. Although the CARS technique is straightforward, it is much more complex both experimentally and computationally than spontaneous Raman spectroscopy<sup>39</sup>, another technique also used in this laboratory. For relatively clean steady state flames sufficient signals are generated from the spontaneous Raman effect to determine both temperature and major species concentrations and thus is the preferred choice of techniques. The CARS technique usually exhibits greater advantages when applied to the more hostile environments. Probably the biggest single advantage of CARS is that the signal emerges as a laser-like beam rather than dispersed over  $4\pi$  steradians. Thus one can collect all the signal remotely, without capturing a large amount of unwanted noise signals. In the field of ballistics several types of environments may be suited for probing with the CARS technique. These environments are hot gases exhausting from guns (muzzle flash problem) and propellants burning within a strand burner. The next several paragraphs expand on these possible applications.

<sup>39</sup>J.A. Vanderhoff, R.A. Beyer and A.J. Kotlar, "Laser Raman Spectroscopy of Flames: Temperature and Concentrations in  $CH_4/N_2O$  Flames." Technical Report ARBRL-TR-02388, January 1982.

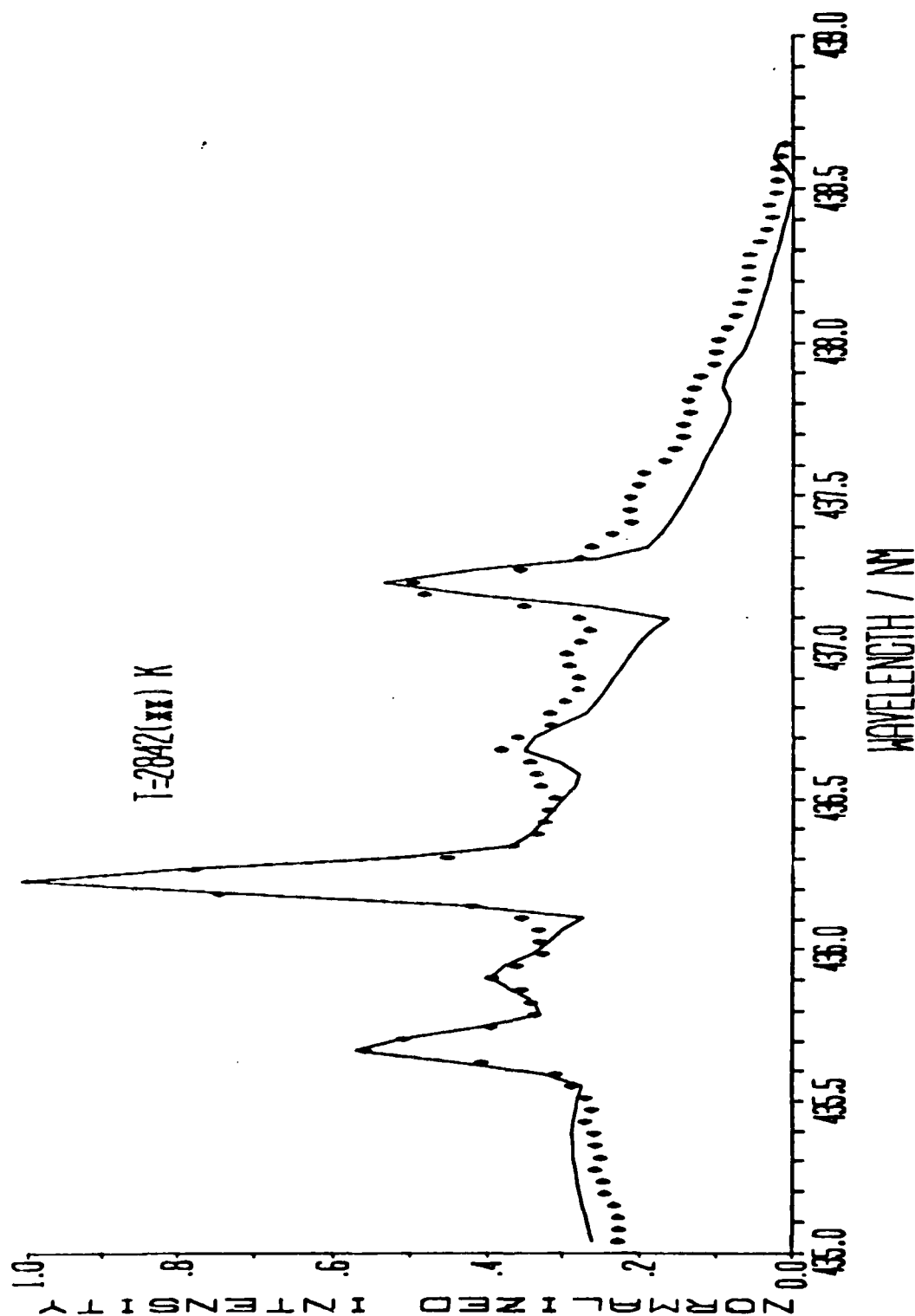


Figure 13. A repeat of the experimental data of Fig. 12d, however in this case an attempt was made to computer fit the spectrum with a non-resonant susceptibility contribution from only within the flame.

Secondary muzzle flash results from the burning of fuel rich exhaust gases with entrained ambient air. Klingenberg<sup>40</sup> has made measurements of temperature, gas velocity, and pressure in secondary flash produced from a 7.62 mm rifle. The time frame he observes is from 1 to 3.4 ms after the exit of the projectile. Here pressures are close to 1 atmosphere and for firing into ambient air the temperatures are in the 2000K range with quoted errors of  $\pm 8\%$  for these sodium and potassium line-reversal methods of temperature determination. The major exhaust gas species are  $H_2O$ ,  $CO$ ,  $CO_2$ ,  $H_2$  and  $N_2$ . The concentration of these species come, almost universally, from the results of thermochemical equilibrium calculations rather than experimental measurements. All of these gases have been previously investigated in a laboratory environment using the CARS technique. Since the pressure, temperature, and composition of gases in the vicinity of muzzle flash are similar to those of the laboratory flames discussed in the text, the CARS signal should be about as large thus amenable for single shot data with a temporal resolution of  $10^{-8}$ s. Additionally, collinear geometry CARS can be used with ease for all of the molecules except  $N_2$  thus simplifying optical alignment while producing substantially more CARS signal than for BOX-CARS geometry. With CARS, single shot temperature measurements to an accuracy of  $\pm 2\%$  should be realizable. Concentrations of the major species can also be obtained from the spectral shape of the CARS signal (see for example Ref. 17). Obtaining concentrations from CARS signals has not been investigated as much as temperature determinations; thus probable error limits are not reported here. One disadvantage of this technique should be pointed out. Although excellent temporal and spatial resolution can be obtained, it would be possible to use only one laser shot per firing. High power Nd:YAG lasers operate at repetition rates on the order of 10 Hz and the time scale of interest for secondary muzzle flash is milliseconds. A high power laser operating at 1 kHz would be very desirable, however, commercial laser technology has not yet advanced to this point.

The other potential application of CARS is to study properties of propellant burning experimentally. Harris and McIlwain<sup>16</sup> have measured the temperature above an unconfined propellant burning in room air using the CARS signal from  $N_2$ . This environment is totally different from the gun barrel environment of interest. An environment which better simulates propellant burning but not as hostile as the gun barrel is a strand burner. The strand burner was designed to measure the burning rate of propellants as a function of pressure. It is a vessel where the pressure, initial temperature, and gas composition are controllable<sup>41</sup>. Two optical ports on this vessel would allow a CARS experiment to be conducted. Temperature and major species concentrations ( $H_2$ ,  $CO$ ,  $N_2$ ,  $H_2O$ ,  $CO_2$ ,  $N_2O$ ,  $NO_2$ ,  $NO$  and  $HCN$ ) could be obtained as a function of various parameters such as pressure, distance from the propellant surface, etc. It must be noted however, that due to probable

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<sup>40</sup>G. Klingenberg, "Investigation of Combustion Phenomena Associated with the Flow of Hot Propellant Gases. III: Experimental Survey of the Formation and Decay of Muzzle Flow Fields and of Pressure Measurements," Combustion and Flame, Vol. 29, pp. 289-309, 1977.

<sup>41</sup>N. Kubota, T.J. Ohlemiller, L.H. Caveny and M. Summerfield, "The Mechanism of Super-Rate Burning of Catalyzed Double Base Propellants," Aerospace and Mechanical Sciences Report Number 1087, 1973, Princeton University, New Jersey.

uneven burning of the propellant surface, distances less than a millimeter from the surface cannot be reliably probed with a double-ended optical technique such as CARS; thus it is not likely that surface temperatures or initial formation of species can be obtained at pressures greater than about one atmosphere when the reaction zone extent is in the submillimeter region. Nonetheless, farther from the propellant surface, temperatures and concentration measurements are feasible and can be compared against current thermochemical equilibrium codes to assess their validity. Multiple data points can be obtained from each strand burning since at atmospheric pressure the propellant burning rate is on the order of 0.1 cm/sec and the region of interest is about 1 cm. Some information may also be obtained at elevated pressure where the CARS signal can increase dramatically. This increase is proportional to various powers of density ( $N$ ) depending on which molecular linewidth regime applies;  $N$  for pressure broadening,  $N^2$  for Doppler broadening, and  $N^3$  for motional narrowing. For further details on the effects see Refs. 38 and 42.

Current efforts<sup>43</sup> are underway to model the high pressure CARS signals for various molecules well enough to extract accurate temperatures and concentrations.

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<sup>42</sup>R.J. Hall, J.F. Verdick and A.C. Eckbreth, "Pressure-Induced Narrowing of the CARS Spectrum of  $N_2$ ," *Opt. Comm.*, Vol. 35, p. 69, 1980.

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